

## INSTITUTE OF ENGINEERING

## Model Entrance Exam

## (Set-11)

## Instructions:

There are 100 multiple-choice questions, each having four choices of which only one choice is correct.

## Section-A (1 marks)

1) A substance has boiling point 563 K , but is starts decomposing near this temperature. Which type of distillation process is suitable for this purification?
a) simple distillation
b) fractional distillation
c) steam distillation
d) distillation under reduced pressure
2) The IUPAC name of $\mathrm{CH}_{3}-\mathrm{O}-\mathrm{C}_{2} \mathrm{H}_{5}$ is:
a) ethoxy methane
b) methoxy ethane
c) ethyl methyl ether
d) methyl ethyl ether
3) Reaction of HBr with propene in absence of peroxide is a/an:
a) electrophilic addition
b) free radical addition
c) electrophilic substitution
d) nucleophilic substitution
4) If an electron has spin quantum number of $+1 / 2$ and a magnetic quantum number of -1 , it cannot be presented in:
a) d-orbital
b) f-orbital
c) p-orbital
d) s-orbital
5) The first step in the extraction of Cu from copper pyrites is:
a) reduction by carbon
b) electrolysis of ore
c) roasting of ore in $\mathrm{O}_{2}$
d) magnetic separation
6) Which of the following is strongest oxidizing agent?
a) $F_{2}$
b) $\mathrm{Cl}_{2}$
c) $B r_{2}$
d) $I_{2}$
7) Copper turnings when heated with concentrated sulphuric acid will give:
a) $\mathrm{SO}_{2}$
b) $\mathrm{SO}_{3}$
c) $\mathrm{H}_{2} \mathrm{~S}$
d) $\mathrm{O}_{2}$
8) Which of the following alkali metal ions has lowest ionic mobility in aqueous solution?
a) $R b^{+}$
b) $\mathrm{Cs}{ }^{+}$
c) $L i^{+}$
d) $\mathrm{Na}{ }^{+}$
9) Which of the following does not reflect the periodicity of elements?
a) Bonding behaviour
b) Electronegativity
c) Ionization Potential
d) Neutron/proton ratio
10) The mass of $112 \mathrm{~cm}^{3}$ of $\mathrm{NH}_{3}$ gas at STP is:
a) 0.085 g
b) 0.85 g
c) 8.5 g
d) 80.5 g
11) Normality of sulphuric acid containing 50 g of the acid in 500 mL of solution is:
a) 2.04 N
b) 0.49 N
c) 0.98 N
d) 0.35 N
12) The organizations $\qquad$ gathered for the seminar on managerial skill improvement.
a) have
b) has
c) was
d) is
13) She $\qquad$ the novel by tomorrow.
a) will be completing
b) has completed
c) had completed
d) will have completed
14) I urged him $\qquad$ the application form.
a) sign
b) to sign
c) signing
d) signed
15) He would rather home tonight than go to the movies.
a) stayed
b) have stayed
c) stay
d) had stayed
16) We must dispense $\qquad$ our duties now. It is getting late.
a) with
b) in
c) on
d) at
17) I was taken by storm when I came to know he passed the exam.
a) felt gloomy
b) felt depressed
c) felt surprised
d) felt foolish
18) Sanchita said to Swastik, "I am planning to leave tomorrow."
a) Sanchita told Swastik that she had planned to leave the next day.
b) Sanchita told Swastik that she is planning to leave the next day.
c) Sanchita told Swastik that she will plan to leave the next day.
d) Sanchita told Swastik that she was planning to leave the next day.
19) Accentuate (Synonym):
a) confuse
b) baffle
c) bewildered
d) emphasize
20) Exquisite (Antonym):
a) hideous
b) elegant
c) dainty
d) delicate
21) The grammatical pattern of the sentence, "Every year, we visit the orphanage." is:
a) $\mathrm{S}+\mathrm{V}+\mathrm{O}+\mathrm{A}$
b) $\mathrm{S}+\mathrm{V}+\mathrm{IO}+\mathrm{DO}$
c) $\mathrm{A}+\mathrm{S}+\mathrm{V}+\mathrm{IO}+\mathrm{DO}$
d) $A+S+V+O$
22) The word 'politician' has a stress primarily on its $\qquad$ syllable.
a) first
b) second
c) third
d) fourth
23) The correct phonetic symbol for the underlined word 'Either' is:
a) $/ \theta /$
b) $/ \int /$
c) $/ \delta /$
d) $/ d /$
24) A matrix $\left(\begin{array}{cc}0 & k+2 \\ 5 & 0\end{array}\right)$ is a skew symmetric matrix if $k=$
a) 3
b) -5
c) -7
d) -2
25) A and $B$ are any two non-empty sets, then $(A \cap \bar{B})=$
a) $A \cup B$
b) $A$
c) $B-A$
d) $A-B$
26) If $f(2 x+1)=x+1$, then $f\left(x^{2}\right)=$
a) $\frac{x^{2}+1}{2}$
b) $\frac{x^{2}+2}{2}$
c) $\frac{x^{2}-1}{2}$
d) $\frac{x^{2}-2}{2}$
27) The sum of 3 numbers of G.P. is 38 and their product is 1728 . Then, the middle term is:
a) 12
b) 8
c) 18
d) 6
28) If $\alpha$ and $\beta$ are the roots of the equation $4 x^{2}+3 x+7=0$, then $\frac{1}{\alpha}+\frac{1}{\beta}=$
a) $\frac{4}{7}$
b) $\frac{-3}{7}$
c) $\frac{7}{3}$
d) $\frac{-7}{4}$
29) If $x=a+b, y=a \omega+b \omega^{2}, z=a \omega^{2}+b \omega$, then the value of $x+y+z=$
a) 0
b) 1
c) -1
d) $\omega^{2}$
30) If $\vec{a}$ and $\vec{b}$ are two non-zero vectors satisfying $|\vec{a}+\vec{b}|=|\vec{a}-\vec{b}|$, then $\vec{a}$ and $\vec{b}$ are:
a) parallel to each other
b) perpendicular to each other
c) inclined at an angle of $60^{\circ}$
d) inclined at an angle of $45^{\circ}$
31) The direction cosines of a line normal to the plane $2 x-y+2 z=0$ is:
a) $2,-1,2$
b) $\frac{2}{9}, \frac{-1}{9}, \frac{2}{9}$
c) $\frac{2}{3}, \frac{-1}{3}, \frac{2}{3}$
d) $1, \frac{-1}{2}, 1$
32) $a x^{2}+2 h x y+b y^{2}=0$ represent a pair of perpendicular lines if:
a) a and b are equal in magnitude but opposite in sign
b) $a$ and $b$ are equal
c) $a$ and $b$ are opposite in sign
d) $a$ and $b$ are reciprocal to each other
33) If latus rectum of an ellipse is half of its minor axis, then its eccentricity is:
a) $\frac{3}{2}$
b) $\frac{2}{3}$
c) $\frac{\sqrt{3}}{2}$
d) $\frac{\sqrt{2}}{3}$
34) If $\vec{a}, \vec{b}$ and $\vec{c}$ are unit vectors such that $\vec{a}+\vec{b}+\vec{c}=0$, then the value of $\vec{a} \cdot \vec{b}+\vec{b} \cdot \vec{c}+\vec{c} \cdot \vec{a}=$
a) 1
b) $3 / 2$
c) $-3 / 2$
d) -3
35) The equation of tangent to the parabola $4 y^{2}+6 x=8 y+7$ at its vertex is:
a) $6 x=11$
b) $6 x+11=0$
c) $y=1$
d) $y+1=0$
36) If $2 \sec 2 \alpha=\tan \beta+\cot \beta$, then the value of $(\alpha+\beta)=$
a) $\frac{\pi}{2}$
b) $\frac{\pi}{3}$
c) $\frac{\pi}{4}$
d) $\frac{\pi}{6}$
37) In $\triangle \mathrm{ABC}, a=2 b$ and $A=3 B$, then angle $A$ equals:
a) $30^{\circ}$
b) $60^{\circ}$
c) $90^{\circ}$
d) $120^{\circ}$
38) If the function $f: R \rightarrow R$ is defined by $f(x)=2 x+\cos x$, then $f$ is:
a) has a minimum at $x=\pi$
b) has a maximum at $x=0$
c) is a decreasing function
d) is an increasing function $\int_{0}^{1} \frac{1-x}{1+x} d x$ equals
a) $2 \log 2-1$
b) $1-2 \log 2$
c) $\sqrt{2} \log 2-1$
d) $2 \log 2+1$
39) $\int \tan ^{-1} \sqrt{\frac{1-\cos 2 x}{1+\cos 2 x}} d x=$
a) $x+c$
b) $\frac{x^{2}}{2}+c$
c) $-x+c$
d) $\frac{-x^{2}}{2}+c$
40) If $y=a \sin m x+b \cos m x$, then $\frac{d^{2} y}{d x^{2}}$ is equal to:
a) $m^{2} y$
b) $-m^{2} y$
c) $m y$
d) $-m y$
41) $\lim _{x \rightarrow \frac{\pi}{4}}\left(\frac{1-\tan x}{1-\sqrt{2} \sin x}\right)=$
a) -2
b) 2
c) -1
d) 0
42) If the function $f(x)=\frac{\sin 3 x}{x}(x \neq 0)=\frac{k}{2}$ when $x=0$ is continuous at $x=0$, then $k=$
a) 2
b) 4
c) 6
d) 3
43) In Joule's law of heating, heat produced is directly proportional to:
a) current
b) square root of current
c) square of current
d) independent of current
44) Internal energy of real gas depends on:
a) Volume
b) Temperature
c) Pressure
d) Volume and Temperature
45) The dimension of Planck constant is same as that of:
a) Angular momentum
b) Linear momentum
c) Energy
d) Resistance
46) If an object of mass ' $m$ ' is moving in a circular path with uniform speed ' $v$ ', which of the following changes occurs in half revolution?
a) kinetic energy changes by $\frac{m v^{2}}{2}$
b) kinetic energy changes by $m v^{2}$
c) momentum changes by 2 mv
d) momentum does not changes
47) The time taken to move from mean position to half of amplitude in SHM is:
a) $\frac{T}{6}$
b) $\frac{T}{12}$
c) $\frac{T}{3}$
d) $\frac{T}{9}$
48) An excess electron is on a spherical body of area $2 \mathrm{~mm}^{2}$. The surface charge density is:
a) $4 \times 10^{-12}$
b) $8 \times 10^{-12}$
c) $4 \times 10^{-14}$
d) $8 \times 10^{-14}$
49) An electric dipole is kept in a uniform electric field. It experiences:
a) a force and a torque
b) a torque but no force
c) a force but not torque
d) neither force nor torque
50) Velocity of a light in vacuum depends upon:
a) frequency
b) wavelength
c) amplitude
d) none of these
51) To detect the obstacles in their path, bats produce:
a) infrasonic waves
b) ultrasonic waves
c) radio waves
d) gamma waves
52) The phenomenon only associated with transverse wave is:
a) reflection
b) refraction
c) polarization
d) interference
53) "Water proofing" agent changes the angle of contact from:
a) obtuse to acute angle
b) acute to obtuse angle
c) obtuse to $90^{\circ}$
d) acute to $90^{\circ}$
54) A coil with its horizontal axis is perpendicular to the magnetic field. The angle between magnetic field and the plane of coil when induced emf is maximum is:
a) $90^{\circ}$
b) $45^{\circ}$
c) $30^{\circ}$
d) $0^{\circ}$
55) A girl presses her physics text book against a rough vertical wall with her hand. The direction of frictional force on the book exerted by the wall is:
a) downwards
b) upwards
c) out from the wall
d) into the wall
56) A PN junction diode can be used as:
a) rectifier
b) capacitor
c) inductor
d) impedance
57) In $\beta$-emission from a radioactive substance, an electron is ejected. This electron comes from:
a) the outermost orbit of an atom
b) the innermost orbits of an atom
c) the surface of a substance
d) the nucleus of an atom
58) A lens behaves as a converging lens in air and diverging lens in water. The refractive index of material of lens is:
a) equal to air
b) equal to water
c) more than air and less than water
d) more than water
59) The velocity of photo electrons emitted in photoelectric effect depends on:
a) wavelength of incident light
b) intensity of incident light
c) photoelectric current
d) both b and c

## Section-B (2 marks)

Read the following passages and answer the questions given below (61-64):
Photosynthesis is the process that plants use to convert sunlight into the food that they need to survive and grow. Most plants create some form of sugar from the sunlight, and this sugar is used by the plant as its primary food source

Plants actually need only three things to create this sugar: sunlight, carbon dioxide, and water. The sunlight reacts with the plant's chlorophyll, a green chemical which is used to convert water and carbon dioxide into sugar.

As a general rule, photosynthesis occurs in a plant's leaves. The leaf contains chlorophyll, which reacts when sunlight strikes the leaf. It is also the chlorophyll which gives the leaf its typical green color, since photosynthesis absorbs most light rays except green, which are reflected outwards.

The process of photosynthesis produces more than just sugar, however. One byproduct of the process is oxygen, which is "exhaled" by the plant into the atmosphere. In fact, plant photosynthesis is one of the primary sources of oxygen generation on our planet, making plant life essential to almost all living things on earth.
61) According to paragraph 4 , which of the following is a byproduct of photosynthesis?
a) green pigment
b) water
c) carbon dioxide
d) oxygen
62) The underlined word essential, as used in paragraph 4 , most nearly means
a) necessary
b) optional
c) fragrant
d) growing
63) The passage explains that photosynthesis is:
a) done in the plant stem.
b) what makes plants edible.
c) the way that plants feed themselves.
d) the source of carbon dioxide in the atmosphere.
64) After reading the passage, what can you infer about photosynthesis?
a) Human life would not survive without plants.
b) Chlorophyll tastes sweet.
c) Sunlight has both good and bad effects on plants.
d) Too much water can interfere with photosynthesis.
65) An element ( X ) which occurs in the second period has an outer electronic configuration $s^{2} p^{1}$, what is the formula and nature of its oxide?
a) $\mathrm{XO}_{3}$, basic
b) $\mathrm{X}_{2} \mathrm{O}_{3}$, basic
c) $\mathrm{XO}_{3}$, acidic
d) $\mathrm{X}_{2} \mathrm{O}_{3}$, acidic
66) The solubility product of a salt having general formula $M X_{2}$ in water is $4 \times 10^{-12}$. The concentration of $M^{2+}$ ions in the aqueous solution of the salt is:
a) $2 \times 10^{-6} \mathrm{M}$
b) $1 \times 10^{-4} \mathrm{M}$
c) $1.6 \times 10^{-4} \mathrm{M}$
d) $4 \times 10^{-10} \mathrm{M}$
67) If $E^{0}{ }_{F e^{2+} / \mathrm{Fe}}=-0.441 \mathrm{~V}$ and $E^{0}{ }_{\mathrm{Fe}^{3+} / \mathrm{Fe}^{2+}}=0.771 \mathrm{~V}$, the standard emf of the reaction $\mathrm{Fe}+2 \mathrm{Fe}^{3+} \longrightarrow 3 \mathrm{Fe}^{2+}$ will be:
a) 1.653 V
b) 1.212 V
c) 0.330 V
d) 0.111 V
68) In the following reaction, $\mathrm{CaCl}_{2} \xrightarrow{\mathrm{H}_{2} \mathrm{O}} P \xrightarrow{\text { hot iron tube }} Q \xrightarrow{\mathrm{CH}_{3} C l, A l C l_{3}} R$, the product ' R ' is:
a) benzene
b) ethyl benzene
c) toluene
d) n-propyl benzene
69) 2 g of metal carbonate is neutralized completely by 100 mL of 0.1 N HCl . The equivalent weight of metal carbonate is:
a) 50
b) 100
c) 150
d) 200
70) The electrons identified by quantum numbers n and l :
(i) $\mathrm{n}=4, \mathrm{l}=1$
(ii) $\mathrm{n}=4, \mathrm{l}=0$
(iii) $\mathrm{n}=3, \mathrm{l}=2$
(iv) $\mathrm{n}=3, \mathrm{l}=1$
can be placed in order of increasing energy as:
a) (iv) < (ii) < (iii) < (i)
b) (ii) < (iv) < (i) < (iii)
c) (i) < (iii) < (ii) < (iv)
d) (iii) < (iv) < (ii) < (i)
71) $\mathrm{MnO}_{2}$ reacts with a halogen acid to give greenish yellow glass. When excess of this gas reacts with $\mathrm{NH}_{3}$, an unstable trihalide is formed. In this process, the oxidation state of nitrogen changes from:
a) -3 to +3
b) -3 to 0
c) -3 to +5
d) 0 to -3
72) The area of triangle formed by the lines $4 x^{2}-9 x y-9 y^{2}=0$ and $x=2$ is:
a) $10 / 3$
b) $20 / 3$
c) 2
d) 3
73) The point of intersection of tangents at the end points of the latus rectum of the parabola $y^{2}=4 x$ is equal to:
a) $(1,0)$
b) $(0,1)$
c) $(-1,0)$
d) $(0,-1)$
74) Equation of a plane passing through (1, $-3,-2$ ) and perpendicular to the planes $x+2 y+2 z=5$ and $3 x+3 y+2 z=8$ is:
a) $2 x+4 y+3 z+16=0$
b) $2 x-4 y-3 z+8=0$
c) $2 x-4 y+3 z-8=0$
d) $2 x+4 y-3 z-16=0$
75) If $\vec{a}, \vec{b}$ and $\vec{c}$ are vectors of magnitude 3,4 and 5 respectively and $\vec{a} \perp(\vec{b}+\vec{c}), \vec{b} \perp(\vec{c}+\vec{a})$ and $\vec{c} \perp$ $(\vec{a}+\vec{b})$, then $|\vec{a}+\vec{b}+\vec{c}|$ is:
a) 50
b) $5 \sqrt{2}$
c) 1
d) 25
76) The lines $2 x-3 y-5=0$ and $3 x-4 y-7=0$ are the diameters of a circle with area $49 \pi$ square units. Then the equation of circle is:
a) $x^{2}+y^{2}+2 x-2 y=67$
b) $x^{2}+y^{2}+2 y-2 x=47$
c) $x^{2}+y^{2}+2 x-2 y=47$
d) $x^{2}+y^{2}-2 x+2 y=62$
77) If $\cos ^{-1} x+\cos ^{-1} y=\frac{\pi}{2}$, and $\tan ^{-1} x-\tan ^{-1} y=0$, then $x^{2}+x y+y^{2}=$
a) 0
b) $\frac{1}{\sqrt{2}}$
c) $\frac{3}{2}$
d) $\frac{1}{8}$
78) The range of $f(x)=\sin ^{-1}\left(\sqrt{x^{2}+x+1}\right)$ is:
a) $\left(0, \frac{\pi}{2}\right]$
b) $\left(0, \frac{\pi}{3}\right]$
c) $\left[\frac{\pi}{2}, \frac{\pi}{3}\right]$
d) $\left[\frac{\pi}{6}, \frac{\pi}{3}\right]$
79) $\left|\begin{array}{ccc}x+4 & x & x \\ x & x+4 & x \\ x & x & x+4\end{array}\right|=$
a) $(x+4)^{3}$
b) $4(x+4)$
c) $16(3 x+4)$
d) $x^{2}(x+4)$
80) If $\mathrm{a}, \mathrm{b}, \mathrm{c}$ are in G.P., $\mathrm{a}, \mathrm{x}, \mathrm{b}$ are in A.P. and $\mathrm{b}, \mathrm{y}, \mathrm{c}$ are in A.P., then $\frac{1}{x}+\frac{1}{y}=$
a) $2 b$
b) $\frac{2}{b}$
c) $a+c$
d) $2 b-a$
81) The number of words which can be formed using letters of the word 'ARRANGE' so that vowels always occupy even places, is:
a) 144
b) 156
c) 72
d) 36
82) $\log _{e} n+\frac{\left(\log _{e} n\right)^{3}}{3!}+\frac{\left(\log _{e} n\right)^{5}}{5!}+\cdots \infty=$
a) $\frac{n^{2}-1}{2 n}$
b) $\frac{n^{2}+1}{2 n}$
c) $\frac{n(n-1)}{2 n}$
d) $\frac{n(n+1)}{2 n}$
83) The value of $\lim _{x \rightarrow 1} \frac{1-\sqrt{x}}{\left(\cos ^{-1} x\right)^{2}}$ is:
a) 4
b) $1 / 2$
c) 2
d) $1 / 4$
84) If $y=\sec \left(\tan ^{-1} x\right)$, then $\frac{d y}{d x}$ at $x=1$ is:
a) $\cos \frac{\pi}{4}$
b) $\sin \frac{\pi}{2}$
c) $\sin \frac{\pi}{6}$
d) $\cos \frac{\pi}{3}$
85) $\int\left(\frac{x+2}{x+4}\right)^{2} e^{x} d x=$
a) $e^{x}\left(\frac{x}{x+4}\right)+c$
b) $e^{x}\left(\frac{x+2}{x+4}\right)+c$
c) $e^{x}\left(\frac{x-2}{x+4}\right)+c$
d) $e^{x}\left(\frac{2 x}{x+4}\right)+c$
86) The area of the rectangle bounded by $|x|=2$, the $x$-axis and $y=1$ (in square units) is:
a) 2
b) 3
c) 4
d) 6
87) A ball is thrown vertically upwards with a velocity of $20 \mathrm{~m} / \mathrm{s}$ from the top of a multistorey building 25 m high. The time taken by the ball to reach the ground is:
a) 2 s
b) 3 s
c) 5 s
d) 7 s
88) The ceiling of a hall is 40 m high. For maximum horizontal distance, the angle at which the ball may be thrown with a speed of $56 \mathrm{~m} / \mathrm{s}$ without hitting the ceiling of the hall is:
a) $90^{\circ}$
b) $30^{\circ}$
c) $45^{\circ}$
d) $60^{\circ}$
89) A solid cylinder of mass M and radius R rolls without slipping down an inclined plane making an angle $\theta$ with the horizontal. Then, its acceleration is:
a) $\frac{1}{3} g \sin \theta$
b) $\frac{2}{3} g \sin \theta$
c) $\frac{2}{5} g \sin \theta$
d) $\frac{2}{7} g \sin \theta$
90) The escape speed of a body on the Earth's surface is $11.2 \mathrm{~km} / \mathrm{s}$. A body is projected with thrice of this speed. The speed of the body when it escapes the gravitational pull of Earth is:
a) $11.2 \mathrm{~km} / \mathrm{s}$
b) $22.4 \sqrt{2} \mathrm{~km} / \mathrm{s}$
c) $\frac{22.4}{\sqrt{2}} \mathrm{~km} / \mathrm{s}$
d) $22.4 \mathrm{~km} / \mathrm{s}$
91) At what velocity does water emerge from an orifice in a tank in which gauge pressure is $3 \times 10^{5} \mathrm{Nm}^{-2}$ before the flow starts? (density of water $=1000 \mathrm{kgm}^{-3}$ )
a) $24.5 \mathrm{~m} / \mathrm{s}$
b) $14.5 \mathrm{~m} / \mathrm{s}$
c) $34.5 \mathrm{~m} / \mathrm{s}$
d) $44.5 \mathrm{~m} / \mathrm{s}$
92) Two moles of an ideal monoatomic gas occupy a volume 2 V at temperature 300 K . It expands to a volume 4 V adiabatically, then the final temperature of a gas is:
a) 179 K
b) 189 K
c) 199 K
d) 219 K
93) A stretched wire emits a fundamental note of 256 Hz . Keeping the stretching force constant and reducing the length of wire by 10 cm , the frequency becomes 320 Hz , the original length of the wire is:
a) 100 cm
b) 50 cm
c) 400 cm
d) 200 cm
94) Two condensers, one of capacity C and other of capacity $\mathrm{C} / 2$ are connected to a V volt battery, as shown in the figure. The work done in fully charging both the condensers is:

a) $\frac{1}{4} C V^{2}$
b) $\frac{3}{4} C V^{2}$
c) $\frac{1}{2} C V^{2}$
d) $2 C V^{2}$
95) A wire of resistance 12 ohms per meter is bent to form a complete circle of radius 10 cm . The resistance between its two diametrically opposite points, A and B as shown in the figure is:

a) $3 \Omega$
b) $6 \pi \Omega$
c) $6 \Omega$
d) $0.6 \pi \Omega$
96) A solenoid has core of a material with relative permeability 500 and its windings carry a current of 1 A. The number of turns of the solenoid is 500 per metre. The magnetization of the material is:
a) $2.5 \times 10^{3} \mathrm{Am}^{-1}$
b) $2.5 \times 10^{5} \mathrm{Am}^{-1}$
c) $2.0 \times 10^{3} \mathrm{Am}^{-1}$
d) $2.0 \times 10^{5} \mathrm{Am}^{-1}$
97) An alternating supply of 220 V is applied across a circuit with resistance $22 \Omega$ and impedance $44 \Omega$. The power dissipated in the circuit is:
a) 1100 W
b) 550 W
c) 2200 W
d) $(2200 / 3) \mathrm{W}$
98) In a compound microscope, the focal length of two lenses are 1.5 cm and 6.25 cm . If an object is placed at 2 cm from objective and the final image is formed at 25 cm from the eye lens. The distance between the two lenses is:
a) 6 cm
b) 7.75 cm
c) 9.25 cm
d) 11 cm
99) In Young's double slit experiment, distance between two sources is 0.1 mm . The distance of screen from the source is 20 cm . Wavelength of light used is $5460 \AA$. Then, angular position of the first dark fringe is:
a) $0.08^{\circ}$
b) $0.16^{\circ}$
c) $0.20^{\circ}$
d) $0.31^{\circ}$
100) The energy required to excite an electron in hydrogen atom to its first excited state is:
a) 8.5 eV
b) 10.2 eV
c) 12.7 eV
d) 13.6 eV

