BEATS ENGINEERING

INSTITUTE OF ENGINEERING

Model Entrance Exam

(Set-13 Solutions)

Instructions:

There are 100 multiple-choice questions, each having four choices of which only one choice is correct.

Date: 2080/05/09 (August-26) Duration: 2 hours Time: 8 AM – 10 AM Section-A (1 marks)

1) c 2) d 3) d 4) a 5) a 6) d 7) a 8) c 9) b 10) a 11) b 12) c 13) c From triangle of vector, $\overrightarrow{AB} + \overrightarrow{BC} + \overrightarrow{CA} = 0$ $\left|\overrightarrow{AB} + \overrightarrow{BC} + \overrightarrow{CA}\right|^2 = 0$ $(AB)^{2} + (BC)^{2} + (CA)^{2} + 2(\overrightarrow{AB}.\overrightarrow{BC} + \overrightarrow{BC}.\overrightarrow{CA} + \overrightarrow{CA}.\overrightarrow{AB}) = 0$ $a^{2} + a^{2} + a^{2} + 2(\overrightarrow{AB}.\overrightarrow{BC} + \overrightarrow{BC}.\overrightarrow{CA} + \overrightarrow{CA}.\overrightarrow{AB}) = 0$ $\overrightarrow{AB}.\overrightarrow{BC} + \overrightarrow{BC}.\overrightarrow{CA} + \overrightarrow{CA}.\overrightarrow{AB} = \frac{-3a^2}{2}$ $x^2 + y^2 = 0$ 14) c $\Rightarrow x = 0 \& y = 0$, which represents z-axis. For both conics, centre = (0,0) and vertices = $(\pm 5, 0)$. 'e' is different in both cases. So, foci and directrices are different. 15) c As, $2x + y + \lambda = 0$ 16) c $y = -2x - \lambda$ $\therefore m = -2, a = -2, c = \lambda$ For normal, $c = 2am - am^3$ $-\lambda = -2(-2)(-2) + 2(-2)^3$ $\lambda = 24$ Put $x = r \cos \theta$, $y = r \sin \theta$ 17) b $r^2 = x^2 + y^2$ $x^{2} + y^{2} + 4x - 6y - 2 = 0$ Centre = (-2, 3)18) b y = mx + cTwo parameters m and c should be known. 19) b Here, n(S) = 49Favorable numbers are 11, 21, 31, 41 \therefore Required probability = $\frac{4}{49}$ On squaring, we get, 20) a $1 + 2\sin\theta \cdot \cos\theta = 1 + 2\sin 2\theta \cdot \cos 2\theta$ $\sin 2\theta = 2\sin 2\theta \,.\, \cos 2\theta$ $\cos 2\theta = \frac{1}{2}$ $\cos 2\theta = \cos \frac{\pi}{2}$ $2\theta = \frac{\pi}{3}$ $\theta = \frac{\pi}{6}$ $\cos^{-1}\frac{1}{x} + \cos^{-1}\frac{1}{y} = \sec^{-1}x + \sec^{-1}y = \csc^{-1}y + \sec^{-1}y = \frac{\pi}{2}$ 21) c $\lim_{x \to 0} \frac{e^{x} - e^{-x}}{\sin x} \qquad \left[\frac{0}{0} \text{ form}\right]$ 22) d

 $= \lim_{x \to 0} \frac{e^{x} - e^{-x}}{\cos x} = \frac{e^{0} + e^{-0}}{\cos 0} = \frac{1+1}{1} = 2$ At x = 2, f(x) is undefined, f(x) is discontinuous at x = 2. 23) b $v = e^{\sqrt{2x}}$ 24) a $\frac{dy}{dx} = \frac{d(e^{\sqrt{2x}})}{d(\sqrt{2x})} \cdot \frac{d(\sqrt{2x})}{d(2x)} \cdot \frac{d(2x)}{d(x)} = e^{\sqrt{2x}} \cdot \frac{1}{2} (2x)^{-\frac{1}{2}} \cdot 2 = \frac{e^{\sqrt{2x}}}{\sqrt{2x}}$ $\int \frac{1 + \cos^2 x}{\sin^2 x} dx = \int (\csc^2 x + \cot^2 x) \, dx = \int (\csc^2 x + \csc^2 x - 1) \, dx = \int (2\csc^2 x - 1) \, dx = -2\cot x - x + \cos^2 x + \cos^2 x - 1) \, dx = -2\cot x - x + \cos^2 x + \cos^2 x - 1)$ 25) c $\int_{0}^{a} \frac{dx}{a^{2} + x^{2}} = \left[\frac{1}{a} \tan^{-1}\left(\frac{x}{a}\right)\right]_{0}^{a} = \frac{1}{a} \tan^{-1}\left(\frac{a}{a}\right) = \frac{1}{a} \tan^{-1} 1 = \frac{1}{a} \cdot \frac{\pi}{4} = \frac{\pi}{4a}$ 26) d $y = x^3 + 3x^2 - 9x + 2$ 27) d $y' = 3x^2 + 6x - 9$ y'' = 6x + 6At point of inflection, $y^{\prime\prime} = 0$ 6x + 6 = 0x = -628) d Let $A = \{1\}$ which is non-empty set with least number of elements. Then, its subsets are the set itself $\{1\}$ and empty set {}. So, the least number of subsets of a non-empty set is 2. $f(x) = e^x + 1$ 29) a Range of e^x is $(0, \infty)$ So, Range of $(e^x + 1)$ is $(0 + 1, \infty + 1)$ i.e., $(1, \infty)$. $|A - \lambda I| = 0$ 30) c $\begin{vmatrix} -3 - \lambda & 4 \\ 2 & -1 - \lambda \end{vmatrix} = 0$ $\lambda = 1, -5$ x + iy = a - ib31) d Taking conjugate on both sides, x - iy = a + ib $x^2 - (p+1)x + pq = 0$ 32) b Sum of the roots = p + 1i.e., p + q = p + 1q = 1 33) a 34) d 35) b Difference between apparent and real depth of a pond is due to refraction. Other three are due to total internal reflection. 36) b 37) d When an ac voltage of 220 V is applied to a capacitor C, the charge on the plates is in phase with the applied voltage. As the circuit is pure capacitive so, the current developed leads the applied voltage by a phase angle of 90° . Hence, power delivered to the capacitor per cycle is: $P = V_{rms} I_{rms} \cos 90^\circ = 0$ An emf is induced only when magnetic flux linked with the loop changes. This is possible when the loop is rotated about 38) b a diameter. 39) c 40) b When the string vibrates in loops n, its frequency is: $f_n = \frac{nv}{2L}$ Where, L is the length of the string and v is the velocity of the wave. : When the string fixed at its both ends vibrate in 1 loop, 2 loops, 3 loops and 4 loops, the frequencies are in the ratio 1:2:3:4

41) a When the lift is at rest, the frequency of oscillation of simple pendulum is, $f = \frac{1}{2\pi} \sqrt{\frac{g}{L}}$.

When the lift falls freely, then effective value of acceleration due to gravity is: g' = g - a = g - g = 0.

$$\therefore f = \frac{1}{2\pi} \sqrt{\frac{0}{L}} = 0$$

42) a $\eta = 1 - \frac{T_2}{T_1} = \frac{T_1 - T_2}{T_1}$

When T_1 and T_2 both are decreased by 100 K each, $(T_1 - T_2)$ stays constant. T_1 decreases.

 $\therefore \eta$ increases.

43) b As,
$$\beta = 2\alpha, \gamma = 3\alpha;$$

 $\frac{\beta}{\gamma} = \frac{2\alpha}{3\alpha} = \frac{2}{3}$

- 44) c When a body is just floating in a liquid whose density is equal to the density of body, is pushed down slightly, then downward force on the body increases due to atmospheric pressure and due to water column above the body. As a result of which, the body sinks in the liquid.
- 45) d Linear momentum is not conserved.
- 46) a As, $x \propto t^3$, $v \propto 3t^2$, $a \propto 6t$ (increasing with time)
- 47) b When water freezes to form ice, the randomness decreases, i.e., entropy decreases.
- 48) d In face-centered cubic lattice, the structure possesses 8 corner atoms and 6 atoms at the centre of each face. Thus, the total number of metal atoms per unit cell in face-centered cubic lattice = $8 \times \frac{1}{8} + 6 \times \frac{1}{2} = 4$
- 49) c 106 g Na₂CO₃ contains $3 \times 6.023 \times 10^{23}$ oxygen atoms.

10.6 g Na₂CO₃ contains $\frac{3 \times 6.023 \times 10^{23}}{106} \times 10.6$ oxygen atoms = 1.806×10^{23} oxygen atoms

- 50) d Maximum number of electrons in a sub-shell = 2(2l + 1) = 2(2.3 + 1) = 14
- 51) c Bond angle decreases with decrease in electronegativity of the central atom.
- 52) c Hydrogen cannot be prepared by the action of dil. H₂SO₄ on copper and mercury as these two metals cannot displace hydrogen from acids. Action of dil. H₂SO₄ on Pb stops after sometime due to the formation of insoluble PbSO₄. Only iron reacts rapidly with dil. H₂SO₄ to give H₂.
- 53) a 54) c

At cathode, $2 H_2O + 2e^- \rightarrow H_2 + 2OH^ Na^+ + OH^- \rightarrow NaOH$ At anode, $Br^- \rightarrow Br + e^ Br + Br \rightarrow Br_2$

- 55) b
- 56) b
- 57) d If an organic compound contains S, in addition to N, then NaCNS is formed which reacts with FeCl₃ to form red coloured ferric sulphocyanide.

 $Na + C + N + S \rightarrow NaCNS$ $3NaCNS + FeCl_3 \rightarrow Fe(CNS)_3 + 3NaCl$

- 58) b
- 59) b All aldehydes (aliphatic and aromatic) and ketones (aliphatic methyl ketones) gives sodium bisulphite test.
- 60) c Order of basicity: $2^{\circ} > 1^{\circ} > 3^{\circ} > NH_3$

Section-B (2 marks)

- 61) a The examples in this passage are mainly about Roosevelt's accomplishments.
- 62) c The second sentence of the first paragraph supports this choice.
- 63) b In the second paragraph, the first sentence supports this answer.
- 64) a This is the only choice and is stated in paragraph 1.
- 65) a Acute angle between the line $x^2 + 4yx + y^2 = 0$ is $\tan^{-1}\left(\frac{2\sqrt{4-1}}{1+1}\right) = \tan^{-1}\sqrt{3} = \frac{\pi}{3}$ Angle bisector of $x^2 + 4yx + y^2$ are:

Angle bisector of
$$x^2 + 4yx + y^2$$

$$2(x^2 - y^2) = (1 - 1)xy$$

$$x^2 = y^2$$

 $x = \pm y$

As, x + y = 0 is perpendicular to x - y = 4, the given triangle is isosceles with vertical angle equal to $\pi/3$ and hence equilateral.

66) b Centre = (0, 10)Radius = $\sqrt{100 - 90} = \sqrt{10}$ Since the line lies outside the circle, p > a $\left|\frac{0-10}{\sqrt{1+m^2}}\right| > \sqrt{10}$ $\frac{10}{\sqrt{1+m^2}} > \sqrt{10}$ $100 > 10 + 10m^2$ $90 > 10m^2$ $9 > m^2$ |m| < 367) b Vertex = (0, 0) $4a = 12 \Rightarrow a = 3$ Ends of latus rectum = $(\pm 2a, a) = (6, 3) \& (-6, 3)$: Required area $=\frac{1}{2}\begin{vmatrix} 0 & 0 & 1 \\ 6 & 3 & 1 \\ -6 & 3 & 1 \end{vmatrix} = 18 \text{ sq. units}$ Here, $l = \frac{-3}{\sqrt{(-3)^2 + (2)^2 + (6)^2}} = \frac{-3}{7}$ $m = \frac{2}{\sqrt{(-3)^2 + (2)^2 + (6)^2}} = \frac{2}{7}$ $n = \frac{6}{\sqrt{(-3)^2 + (2)^2 + (6)^2}} = \frac{6}{7}$ 68) d p = 7Hence, equation of plane in normal form will be: $-\frac{3}{2}x + \frac{2}{7}x + \frac{6}{7}z = 7$ -3x + 2y + 6z - 49 = 0 \vec{a} lies in the plane of \vec{b} and $\vec{c} \Rightarrow \vec{a}, \vec{b}, \vec{c}$ are coplanar. 69) c $\left[\vec{a}\ \vec{b}\ \vec{c}\right] = 0$ $\Rightarrow \vec{a}.(\vec{b} \times \vec{c}) = 0$ $\bar{X} = \frac{\sum x_i}{n}$ 70) a $\sum x_i = 36 \times 50 = 1800$ Now, $\sum x_i - (42 + 30) = 1800 - 72 = 1728$: New mean $=\frac{1728}{48}=36$ $2(r+R) = 2\left[(s-c)\tan\frac{c}{2} + \frac{c}{2\sin c}\right]$ 71) a $= 2\left[(s-c) + \frac{c}{2}\right] \qquad \left(\because C = \frac{\pi}{2}\right)$ = 2s - 2c + c $= a + b + c - c \qquad (\because 2s = a + b + c)$ = a + b72) d $\frac{C_{r-1}^{20}}{C_r^{20}} = \frac{1}{2}$ $\frac{20!}{(r-1)!(21-r)!} \times \frac{r!(20-r)!}{20!}$ $\frac{1}{2} = \frac{r}{21-r}$ r = 7Given, $T_n = \frac{1}{2} \left[\frac{3^n}{n!} - \frac{1}{n!} \right]$ 73) b

Sum = $\sum_{n=1}^{\infty} T_n = \frac{1}{2} \left[\sum_{n=1}^{\infty} \frac{3^n}{n!} - \sum_{n=1}^{\infty} \frac{1}{n!} \right]$

$$= \frac{1}{2} \left[\left\{ 1 + \frac{3}{11} + \frac{3^2}{21} + \cdots \right\} \left\{ 1 + \frac{1}{11} + \frac{1}{21} + \cdots \right\} \right]$$

$$= \frac{1}{2} (e^3 - e)$$
74) a
$$\boxed{\begin{array}{c|c|c|c|c|} \hline Ladies (4) & Gentlemen(6) & Number of Selection \\\hline 2 & 3 & 4C_2 \times {}^{6}C_3 = 120 \\\hline 3 & 2 & 4C_3 \times {}^{6}C_2 = 60 \\\hline \hline 4 & 1 & 4C_4 \times {}^{6}C_1 = 6 \\\hline \end{array}}$$
Total = 120 + 60 + 6 = 186
75) d
$$f(2^x) = g(x^2)$$

$$(2^x)^2 = 2^{x^2}$$
Applying log on both sides,
$$2x \log 2 = x^2 \log 2$$

$$2x - x^2 = 0$$

$$x(2 - x) = 0$$

$$x = 0, 2$$
Solution set = {0, 2}
76) c
$$3 + 3a + 3a^2 + \cdots = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{45}{8}$$

$$3(1 + a + a^2 + \cdots = 2) = \frac{2}{8} = \frac{2}{12} = \frac{2}{1$$

81) b As the number of undecayed nuclei decreases from 25% to 12.5% in 10 s, it shows that the half-life of the sample is 10 s. i.e., $T_{1/2} = 10 s$

Decay constant, $\lambda = \frac{0.693}{T_{1/2}} = \frac{0.693}{10}$ Mean life, $\tau = \frac{1}{\lambda} = \frac{10}{0.693} = 14.43 \text{ s}$

BEATS

82) b
$$\frac{1}{4\pi} = \frac{1}{|2\pi|} \frac{1}{|2\pi|} = \frac{1}{2\pi} \times \frac{\pi}{2} = \frac{\pi}{2\pi}$$

 $\lambda = (2Q/2)\lambda_{0}$
83) b As reflected light is completely polarized, therefore, $l_{p} = 60^{\circ}$
 $\mu = \tan l_{p} = \tan 60^{\circ} = \sqrt{3}$
 $A = \pi l_{p} = \frac{1}{2\pi} \frac{1}{2\pi} = \sqrt{3} \times 10^{\circ} \text{ m/s}$
84) a For minimum deviation, $t = e$
 $r_{1} = r_{2} = \frac{1}{2\pi} \frac{e^{-2}}{2} = 30^{\circ}$
Using Smell's have at the interface,
 $1 \sin t = \sqrt{3} \sin r_{1}$
 $\sin t = \sqrt{3} \sin r_{1}$
 $\pi = \frac{1}{\pi \sin r^{2}}}$
 $\pi = \frac{1}{\pi \sin r^{2}}}$

$$a = \frac{g\sin\theta}{1 + \frac{I}{MR^2}}$$

For a solid sphere, $I = \frac{2}{5}MR^2$

$$a = \frac{g\sin\theta}{1+\frac{2}{5}} = \frac{5}{7} g\sin\theta$$

Acceleration of the same sphere when it slides without friction down an same inclined plane is

$$a' = g \sin \theta$$

So, $\frac{a'}{a} = \frac{\frac{5}{7}g \sin \theta}{g \sin \theta} = \frac{7}{5}$
 $a' = \frac{7}{5}a$

92) d Since ranges are same so shell have been fired at complementary angles. Let, they be θ and $90^{\circ} - \theta$. $t = -\frac{2u \sin \theta}{2u \sin \theta}$ and $t = -\frac{2u \sin(90^{\circ} - \theta)}{2u \cos \theta} - \frac{2u \cos \theta}{2u \cos \theta}$

$$\begin{array}{l} t_{1} = \frac{1}{g} - \inf_{g} t_{1} t_{2} = -\frac{1}{g} - \frac{1}{g} - \frac{1}{g} \\ t_{1} t_{2} = \frac{2}{g} \cdot \frac{2u^{2} \sin \theta \cos \theta}{g} = \frac{2}{g} \cdot \frac{u^{2} \sin 2\theta}{g} = \frac{2R}{g} \quad \left[\because R = \frac{u^{2} \sin 2\theta}{g} \right] \\ \end{array} \\ \begin{array}{l} \text{93) a} & \text{Mg} + \frac{1}{2} O_{2} \rightarrow \text{MgO} \\ 24 \text{ g Mg react with 16 g of } O_{2} \\ 1 \text{ g Mg reacts with } \frac{16}{24} = \frac{2}{3} = 0.67 \text{ g of } O_{2} \\ \text{However, only 0.56 g of } O_{2} \text{ is present, thus Mg is present in excess.} \\ 0.67 \text{ g of } O_{2} \text{ reacts with 1 g of Mg} \\ 0.56 \text{ g of } O_{2} \text{ reacts with 1 g of Mg} \\ 0.56 \text{ g of } O_{2} \text{ with } \frac{1}{0.67} \times 0.56 = \frac{56}{67} = 0.84 \text{ g Mg} \\ \text{Mg left = 1 - 0.84 = 0.16 g} \\ 94) \text{ b} & 2MnO_{4}^{-} + Br^{-} + H_{2}O \rightarrow 2MnO_{2} + BrO_{3}^{-} + 2OH^{-} \\ +7 & +4 \\ 95) \text{ b} & K_{c} = \frac{[NO_{2}]^{2}}{[N_{2}O_{4}]} = \frac{(1.2 \times 10^{-2})^{2}}{4.8 \times 10^{-2}} = 3 \times 10^{-3} \text{ mol } L^{-1} \\ 96) \text{ a} & W = ZIt \\ It = \frac{W}{Z} = 1 & \because (W = Z) \\ t = \frac{1}{l} = \frac{1}{0.25} = 4 \text{ A} \\ 97) \text{ b} & 3Mg + N_{2} \rightarrow Mg_{3}N_{2} \\ Mg_{3}N_{2} + H_{2}O \rightarrow Mg(OH)_{2} + NH_{3} \\ CuSO_{4} + 4NH_{3} \rightarrow Cu[(NH_{3})_{4}]SO_{4} \\ \end{array}$$

- 98) a Sulphur trioxide is prepared by heating conc. H_2SO_4 with large excess of phosphorous pentoxide, P_2O_5 . $H_2SO_4 + P_2O_5 \rightarrow SO_3 + 2HPO_3$
- 99) a

Thank You!!!!!!