



INSTITUTE OF ENGINEERING

MODEL ENTRANCE EXAM

(SET – 6)

Instructions:

There are 100 multiple-choice questions, each having four choices of which only one choice is correct.

Date : 2081/03/15
(June 29)

Duration : 2 hours
Time : 8 A.M. – 10 A.M.

- 18) If $x \in [0, 2\pi]$, then the solution set of the inequation $4\sin^2 x - 8\sin x + 3 \leq 0$, is:
a) $[0, \pi/6]$ b) $[0, 5\pi/6]$ c) $[5\pi/6, 2\pi]$ d) $[\pi/6, 5\pi/6]$
- 19) Family of curves $y = Ax + A^3$ is represented by the differential equation of degree:
a) 3 b) 2 c) 1 d) 0
- 20) The arithmetic mean of ${}^n C_0, {}^n C_1, \dots, {}^n C_n$ is:
a) $\frac{1}{n}$ b) $\frac{2^n}{n}$ c) $\frac{2^{n-1}}{n}$ d) $\frac{2^{n+1}}{n}$
- 21) The oxidation number of Iron in $[\text{Fe}(\text{H}_2\text{O})_5\text{NO}]\text{SO}_4$ is:
a) 1 b) 2 c) 3 d) 0
- 22) Which of the following contains both covalent and co-ordinate bond?
a) CO b) CO_2 c) CaCl_2 d) C_2H_6
- 23) The substance which causes permanent hardness in water is:
a) NaCl b) NaHCO_3 c) MgCl_2 d) K_2SO_4
- 24) When concentrated H_2SO_4 is added to dry KNO_3 , brown fumes evolve. These fumes are of:
a) SO_2 b) SO_3 c) NO_2 d) NO
- 25) In order to prevent the hot metal filament from getting burnt, when the electric current is switched on, the bulb is filled with:
a) Cl_2 b) H_2 c) NH_3 d) an inert gas
- 26) The purpose of smelting an ore is to:
a) reduce the ore b) oxidize the ore
c) obtain an alloy d) separate volatile impurities
- 27) Excess of NaOH reacts with zinc to form:
a) $\text{Zn}(\text{OH})_2$ b) ZnO c) ZnH_2 d) Na_2ZnO_2
- 28) A signature written with carbon pencil weighs 1mg. What is the number of carbon atoms present in the signature?
a) 6.02×10^{20} b) 0.502×10^{20} c) 5.62×10^{23} d) 5.02×10^{20}
- 29) Which of the following represent correct set of quantum number of 4d electron?
a) 4, 3, 2, +1/2 b) 4, 2, 1, 0 c) 4, 3, -2, +1/2 d) 4, 2, 1, -1/2
- 30) In face centered arrangement, the number of atoms per unit cell is:
a) 8 b) 2 c) 1 d) 4
- 31) Which among the following is not a state function?
a) Internal energy b) Free energy c) Work d) Enthalpy
- 32) A liquid decomposes at its boiling point. It can be purified by:
a) sublimation b) steam distillation
c) vacuum distillation d) fractional distillation
- 33) Chloroform on warming with Ag powder, gives:
a) C_2H_6 b) C_3H_6 c) C_2H_4 d) C_2H_2
- 34) Formalin is 40% aqueous solution of:
a) methanoic acid b) methanal c) methanol d) methanamine
- 35) The dimensions of physical quantity X in the equation $\text{Force} = \frac{X}{\text{Density}}$ is given by:
a) $\text{M}^1\text{L}^4\text{T}^{-2}$ b) $\text{M}^2\text{L}^{-2}\text{T}^{-1}$ c) $\text{M}^2\text{L}^{-2}\text{T}^{-2}$ d) $\text{M}^1\text{L}^{-2}\text{T}^{-1}$
- 36) If action and reaction forces are always equal in magnitude, then these forces:
a) will produce accelerations of equal magnitudes
b) may not produce accelerations of equal magnitudes
c) produce velocities of equal magnitudes
d) will not produce accelerations of equal magnitudes
- 37) The velocity of centre of mass of the system remains constant, if the total external force acting on the system is:
a) minimum b) maximum c) unity d) zero

